



ANTIMICROBIAL POTENT OXAZINE-2-AMINE DERIVATIVES

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ABSTRACT

Objective: To synthesize thirteen effective microbial potent oxazine-2-amine derivatives and evaluate the antimicrobial activities of these compounds. **Material and methods:** The solvent-free solid fly-ash: H_2SO_4 catalyzed cyclization of 4-bromo-1-naphthyl chalcones and urea under microwave irradiation technique was utilized for the synthesis of oxazine derivatives. The synthesized oxazines were characterized by their physical constants and spectroscopic data. The Bauer-Kirby disc-diffusion method of measurement of zone of inhibition was used for evaluation antimicrobial activity of synthesized oxazines. **Results:** The cyclization was effective and this reaction gave more than 80% yield. The catalyst was reused up to five times consecutively without appreciable decreasing of yield. All compounds were active against their microbial strains. **Conclusions:** This solvent-free synthetic method was effective for the synthesis of oxazine derivatives. The amino, dimethylamino, halogens, methoxy and methyl substituted oxazines shows significant antibacterial activity. Chloro, hydroxy and nitro substituted oxazine derivative have shown significant antifungal activity.

Keywords: Microwave-assisted synthesis; oxazines, IR and NMR spectra, antibacterial activity, antifungal activity.

INTRODUCTION

The unsaturated six-membered heterocyclic compounds possess one oxygen and one nitrogen atoms are known as oxazines [1-4]. These oxazine molecules exist many isomeric conformers such as 1, 2 or 1, 3 or 1, 4 oxazines types[5] depend upon the relative position of these two atoms and the double bond. These oxazines were biologically important due to the presence of oxygen, nitrogen heteroatoms along with double bonds in their structural moieties [6]. The important biological activities of these oxazine derivatives are antibacterial [6-8], anti-fungal [6-8], anti-plasmodial [9], anti-cancer [10], anti-depressants [11], cytotoxicity [12], anti-osteoplastic [13], anti-tumour [14], anti-oxidant [15], anti-tuberculosis [16], anti-neoplastic [17], antagonists [18], anti-inflammatory [19], anti-infectants [20], IKB kinase beta [21] and PTP-1B inhibition [22]. These oxazine derivatives were applied for transcription factor NF-kappaB [23], synthesis of eosinophils [24], identification and separation of neutrophils [25]. Many oxazine derivatives were used as dyes [26]. Reddy et al., have studied the antimicrobial activities of aloe vera extract [27]. The in-vitro antibacterial activity of ethanol, chloroform, hexane and water extracts of leaves and fruits of Aegle marmelos against Methicillin-resistant *Staphylococcus aureus* was reported by Ganapathy and karpagam [28]. Ukaegbu-Obi et. al. have assessed the antimicrobial activities of aqueous and ethanolic extracts of *monodora myristica* (ehuru) seeds [29]. Numerous solvent assisted and solvent-free synthetic methods were available for synthesis of oxazine derivatives [30]. Now-a-days scientists, organic chemists are interested for solvent-free synthesis [7, 31-36]. Some reactions such as hetero Diels-alder reaction [6], ring closure [37], Betti base induced condensation [38, 39], Mannich type condensation-cyclization[7] and cyclization of chalcones[8] were used for the synthesis of oxazine derivatives. Verma et. al. [33] have synthesized some benzoxazine/oxazine fused isoquinolines and naphthyridines by the solvent-free method. Elarfi and Al-difar [6] have synthesized some 1, 3-oxazine derivatives by a solvent-assisted method from chalcones and urea. More than 75% yield of dihydro-²H-benzo- and naphtho-1, 3-oxazine derivatives were prepared by Mathew et al.[5] using the eco-friendly method. Efficient synthesis of some 1, 3-oxazine-4-thiones was synthesized by N-methyl imidazole promoted solvent-free conditions. Sapkal et al., have studied the role of ammonium

acetate for the solvent-free synthesis of 1,3-disubstituted-2,3-dihydro-¹H-naphthyl oxazines [35]. Within the above view, there is no information available in the literature for the solvent-free synthesis the evaluation of biological activities of 4-bromo-1-naphthyl based oxazine-2-amine derivatives. Therefore the author have taken the effort to synthesize some 4-bromo-1-naphthyl based oxazine-2-amines by solvent-free cyclization, characterized by their analytical, spectral data and studied their antimicrobial activities using Bauer-Kirby [40] method.

MATERIAL AND METHODS

General

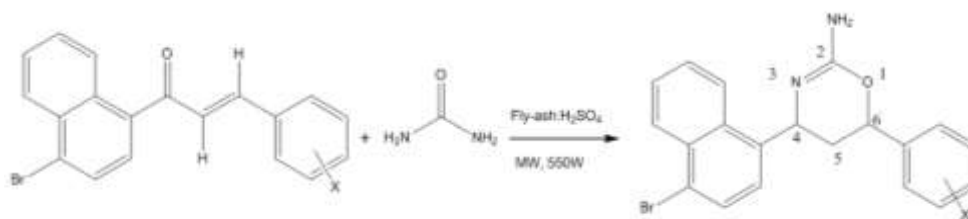
Chemicals were used in this present investigation were purchased from Sigma-Aldrich and Merck Chemical companies. The melting point of oxazine-2-amines were determining in Mettler FP51 apparatus using open glass capillaries and are uncorrected. The infrared spectra of oxazines were recorded in AVATAR-300 Fourier transform spectrophotometer (KBr, 4000-400 cm^{-1}) using KBr disc. The NMR spectra of oxazine-2-amine derivatives were recorded in Bruker AV400 series type NMR spectrometer operating at 400MHz for ¹H and 100 MHz for ¹³C spectra in $CDCl_3$ solvent using TMS as an internal standard. Mass spectra of all synthesized oxazine derivatives were recorded on SHIMADZU mass spectrometer using chemical ionization technique.

Preparation of fly-ash: H_2SO_4 catalyst

The fly-ash: H_2SO_4 catalyst was prepared according to a literature procedure[41].

Synthesis of 4-(aryl)-5,6-dihydro-6-(substituted phenyl)-4H-1,3-oxazine-2-amines.

An equi-molar quantities of chalcones (2 mmol), urea (2mmol) and 0.4 g of fly-ash: H_2SO_4 were subjected to microwave irradiation for 2-4 minutes at 550W (Scheme 1) (Samsung, Microwave Oven, 100-700W). The completion of the reaction was monitored by TLC. After completion of the reaction, dichloromethane (20 mL) was added, followed by simple filtration. The solution was concentrated; the obtained solid was purified by re-crystallization. The solid catalyst was washed with ethyl acetate, dried in hot air oven at 110°C for 1h, and then it is used for further reaction runs.



Entry	1	2	3	4	5	6	7	8	9	10	11	12	13
X	H	3-NH ₂	4-NH ₂	3-Br	3-Cl	4-Cl	4-N(CH ₃) ₂	4-OH	4-OCH ₃	4-CH ₃	2-NO ₂	3-NO ₂	4-NO ₂

Scheme 1: Synthesis of 4-aryl-5,6-dihydro-6-(substituted phenyl)-4H-1,3-oxazine-2-amines by fly-ash:H₂SO₄ catalyzed cyclization of aryl chalcones and urea under microwave irradiation.

Measurement of Antimicrobial activities

Antibacterial activity

The antibacterial activities of all prepared 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines have been evaluated against two gram positive pathogenic strains *Staphylococcus aureus*, *Enterococcus faecalis* and four gram negative strains *Escherichia coli*, *Klebsiella species*, *Pseudomonas*, and *Proteus vulgaris*. The disc diffusion technique was followed using Kirby-Bauer [38] method, at a concentration of 250 µg/mL with Ampicillin and Streptomycin used as the standard drugs.

Antibacterial activities of all 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines were evaluated using Kirby-Bauer [38] disc diffusion technique. In each Petri plate about 0.5 mL of the test bacterial sample was spread uniformly over the solidified Mueller Hinton agar using sterile glass spreader. Then the discs with 5 mm diameter made up of Whatmann No.1 filter paper, impregnated with the solution of the compound were placed on the medium using sterile forceps. The plates were incubated for 24 h at 37°C by keeping the plates upside down to prevent the collection of water droplets over the medium. After 24 h, the plates were visually examined and the diameter values of the zone of inhibition were measured. Triplicate results were recorded by repeating the same procedure.

Antifungal sensitivity assay

The study of antifungal activities of all 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines has been done with *Candida albicans* as the fungal strain using the disc diffusion technique and the other two strains *Penicillium*

species and *Aspergillus niger*, the dilution method was adopted. The drug dilution was kept as 50µg/mL. *Griseofulvin* has been taken as the standard drug.

Antifungal sensitivity assay all 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines were performed using Kirby-Bauer[35] disc diffusion technique. PDA medium was prepared and sterilized as above. It was poured (ear bearing heating condition) in the Petri-plate which was already filled with 1 mL of the fungal species. The plate was rotated clockwise and counter clockwise for uniform spreading of the species. The discs were impregnated with the test solution. The test solution was prepared by dissolving 15 mg of the chalcone in 1mL of DMSO solvent. The medium was allowed to solidify and kept for 24 h. Then the plates were visually examined and the diameter values of the zone of inhibition were measured. Triplicate results were recorded by repeating the same procedure.

RESULTS

The synthesized 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines were characterized by their physical constants and spectral data. Obtained analytical and mass spectral fragments of synthesized oxazines were presented in **Table 1**. Further the compounds were characterized by their infrared and NMR spectra. The obtained spectroscopic data of these oxazines were summarized as follows.

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-phenyl-4H-1,3-oxazine-2-amine (1): FT-IR (KBr): 3427(NH), 1598(C=N), 1253(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.71(s, 2H, NH₂), 5.29(t, 1H, H₄), 2.89 (dd, 1H, H₅), 2.13 (dd, 1H, H₅), 4.78(t, 1H, H₆), 7.03-8.13(m, 11H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 155.19(C₂), 49.75(C₄), 39.83(C₅), 73.19(C₆), 124.39-140.90(Ar-C).

Table 1: Analytical, physical constants, yield and mass fragment of 4-aryl-5,6-dihydro-6(substituted phenyl)-4H-1,3-oxazine-2-amines.

Entry	X	M.W.	Time(m)	Yield(%)	m.p.(°C)	Mass(m/z)
1	H	380	2.5	90	124-125	380[M ⁺], 382[M ²⁺], 364, 305, 303, 301, 286, 204, 175, 127, 99, 84, 77, 58, 43, 42, 16
2	3-NH ₂	396	2	86	103-104	396[M ⁺], 398[M ²⁺], 379, 337, 303, 204 175, 127, 118, 111, 84, 77, 43, 58,16,
3	4-NH ₂	396	2	88	113-115	396[M ⁺], 398[M ²⁺], 379, 337, 204 175, 127, 111, 99, 97, 79, 77, 43, 58,16,
4	3-Br	460	3	85	117-118	460[M ⁺], 462[M ²⁺], 464[M ⁴⁺], 441, 303, 339, 252, 209, 175, 127, 84, 79,77, 58, 43, 35,16
5	3-Cl	415	3	85	99-101	415[M ⁺], 417[M ²⁺], 419[M ⁴⁺], 397, 379, 303, 335, 225, 209, 175, 127, 99, 77, 58, 43, 35,16,
6	4-Cl	415	3	86	131-132	415[M ⁺], 417[M ²⁺], 419[M ⁴⁺], 397, 379, 303, 335, 209, 175, 127, 99, 77, 58, 43,41, 35,16,
7	4-N(CH ₃) ₂	424	3.5	88	124-125	424[M ⁺], 426[M ²⁺], 408, 379, 354, 339,290, 263, 218, 205, 204, 165, 127, 120, 107, 91, 77, 58, 43, 42, 31, 16
8	4-OH	397	2.5	90	118-119	397[M ⁺], 399[M ²⁺], 379, 303, 317, 204, 191, 175, 127, 91, 77, 58, 43, 42, 31, 16
9	4-OCH ₃	411	2	90	102-103	411[M ⁺], 413[M ²⁺], 395, 394, 379, 331,205, 191, 175, 148, 127, 107, 91, 77, 58, 43, 42, 31, 16

10	4-CH ₃	395	2.5	92	146-147	395[M ⁺], 397[M ²⁺], 379, 378, 303, 204, 189, 174, 132, 91, 77, 46, 43, 41, 16,
11	2-NO ₂	426	4	85	116-117	426[M ⁺], 428[M ²⁺], 379, 367, 346, 205, 165, 84,77, 46, 43, 41, 16,
12	3-NO ₂	426	4	85	105-106	426[M ⁺], 428[M ²⁺], 3409, 79, 367, 346, 303,232, 220, 205, 204, 165, 136, 127, 122, 88, 84,79, 77, 58, 46, 45, 43, 41, 29, 16,
13	4-NO ₂	426	4	85	125-126	426[M ⁺], 428[M ²⁺], 340, 379, 367, 346, 220, 205, 204, 175, 165, 136, 127, 122, 88, 84,77, 46, 45, 43, 41, 29, 16,

4-(4-Bromo-1-naphthyl)-5, 6-dihydro-6-(3-aminophenyl)-4H-1, 3-oxazine-2-amine(2): FT-IR (KBr): 3564(NH), 1612(C=N), 1245(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.685(s, 2H, NH₂), 5.32(t, 1H, H₄), 2.46(dd, 1H, H₅), 2.20(dd, 1H, H₅), 4.35(t, 1H, H₆), 6.89-7.25(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.82 (C₂), 51.36 (C₄), 47.98 (C₅), 76.25 (C₆), 126.25-139.38 (Ar-C).

4-(4-Bromo-1-naphthyl)-5, 6-dihydro-6-(4-aminophenyl)-4H-1, 3-oxazine-2-amine (3): FT-IR (KBr): 3556(NH), 1602(C=N), 1212(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.29(s, 2H, NH₂), 4.52 (t, 1H, H₄), 2.12(dd, 1H, H₅), 2.32 (dd, 1H, H₅), 4.87(t, 1H, H₆), 6.86-7.65 (m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.84 (C₂), 51.32 (C₄), 47.93(C₅), 76.54 (C₆), 126.11-139.32 (Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(3-bromophenyl)-4H-1, 3-oxazine-2-amine(4): FT-IR (KBr): 3552(NH), 1623(C=N), 1209(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.22(s, 2H, NH₂), 5.34 (t, 1H, H₄), 2.32(dd, 1H, H₅), 2.36 (dd, 1H, H₅), 4.36(t, 1H, H₆), 6.85-7.28 (m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.34 (C₂), 51.67 (C₄), 47.76(C₅), 72.43 (C₆), 124.26-139.39 (Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(3-chlorophenyl)-4H-1, 3-oxazine-2-amine(5): FT-IR (KBr): 3548(NH), 1614(C=N), 1215(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.21 (s, 2H, NH₂), 5.32(t, 1H, H₄), 2.42(dd, 1H, H₅), 2.65 (dd, 1H, H₅), 4.32(t, 1H, H₆), 6.23-7.27 (m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.33 (C₂), 51.23 (C₄), 47.87(C₅), 66.34 (C₆), 126.23-139.15 (Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(4-chlorophenyl)-4H-1, 3-oxazine-2-amine(6): FT-IR (KBr): 3552(NH), 1610(C=N), 1232(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.27 (s, 2H, NH₂), 5.33 (t, 1H, H₄), 2.42(dd, 1H, H₅), 2.83(dd, 1H, H₅), 4.38(t, 1H, H₆), 6.24-7.67 (m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.38 (C₂), 51.27 (C₄), 47.76(C₅), 66.98 (C₆), 126.37-139.87(Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(4 dimethylamino-phenyl)-4H-1,3-oxazine-2-amine(7): FT-IR (KBr): 3535(NH), 1611(C=N), 1212(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.53 (s, 2H, NH₂), 5.34 (t, 1H, H₄), 2.31(dd, 1H, H₅), 2.73(dd, 1H, H₅), 4.36(t, 1H, H₆), 3.63(s, 6H, N(CH₃)₂), 6.78-7.98 (m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.26 (C₂), 51.27 (C₄), 47.76 (C₅), 66.98 (C₆), 44.87 (N(CH₃)₂), 126.67-139.23(Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(4-hydroxyphenyl)-4H-1, 3-oxazine-2-amine (8): FT-IR (KBr): 3546(NH), 1601(C=N),

1218(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.22 (s, 2H, NH₂), 5.45 (t, 1H, H₄), 2.41(dd, 1H, H₅), 2.62(dd, 1H, H₅), 4.65(t, 1H, H₆), 6.24-7.26(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.34 (C₂), 51.76 (C₄), 47.54(C₅), 66.89 (C₆), 44.19 (N(CH₃)₂), 126.34-139.80(Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(4-methoxyphenyl)-4H-1, 3-oxazine-2-amine (9): FT-IR (KBr): 3540(NH), 1611(C=N), 1223(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.21(s, 2H, NH₂), 5.33 (t, 1H, H₄), 2.44(dd, 1H, H₅), 2.63(dd, 1H, H₅), 4.35(t, 1H, H₆), 4.26 (s, 3H, OCH₃), 6.25-7.37(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.29 (C₂), 51.28 (C₄), 47.55(C₅), 73.68 (C₆), 62.36(OCH₃), 126.32-139.59(Ar-C).

4-(4-Bromo-1-naphthyl)-5,6-dihydro-6-(4-methylphenyl)-4H-1, 3-oxazine-2-amine (10): FT-IR (KBr): 3534(NH), 1610(C=N), 1231(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.22 (s, 2H, NH₂), 5.39 (t, 1H, H₄), 2.47(dd, 1H, H₅), 2.54(dd, 1H, H₅), 4.35(t, 1H, H₆), 2.78 (s, 3H, CH₃), 6.23-7.29(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 154.25 (C₂), 51.36 (C₄), 47.80(C₅), 66.29 (C₆), 26.78(CH₃), 114.98-145.09(Ar-C).

4-(4-Bromo-1-naphthyl)-5, 6-dihydro-6-(2-nitrophenyl)-4H-1, 3-oxazine-2-amine (11): FT-IR (KBr): 3555(NH), 1620(C=N), 1245(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.26 (s, 1H, NH₂), 5.33 (t, 1H, H₄), 2.44(dd, 1H, H₅), 2.64(dd, 1H, H₅), 4.46(t, 1H, H₆), 6.24-7.56(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 156.11 (C₂), 51.29 (C₄), 47.36(C₅), 66.23 (C₆), 114.23-145.25(Ar-C).

4-(4-Bromo-1-naphthyl)-5, 6-dihydro-6-(2-nitrophenyl)-4H-1, 3-oxazine-2-amine(12): FT-IR (KBr): 3550(NH), 1625(C=N), 1230(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.23(s, 1H, NH₂), 5.36 (t, 1H, H₄), 2.45(dd, 1H, H₅), 2.63(dd, 1H, H₅), 4.59(t, 1H, H₆), 6.52-7.90(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 155.69 (C₂), 51.35 (C₄), 47.42(C₅), 66.36 (C₆), 124.90-145.69(Ar-C).

4-(4-Bromo-1-naphthyl)-5, 6-dihydro-6-(2-nitrophenyl)-4H-1,3-oxazine-2-amine(13): FT-IR (KBr): 3545(NH), 1627(C=N), 1235(C-O-C) cm⁻¹. ¹H NMR (CDCl₃-d₆, TMS) δ: 6.22 (s, 1H, NH₂), 5.33 (t, 1H, H₄), 2.27(dd, 1H, H₅), 2.57(dd, 1H, H₅), 4.39(t, 1H, H₆), 6.57-7.89(m, 10H, Ar-H) ppm. ¹³C NMR (CDCl₃-d₆, TMS) δ: 155.70(C₂), 51.38 (C₄), 47.48(C₅), 66.85 (C₆), 124.58-145.90(Ar-C).

Antibacterial activity

The measured antibacterial activity of synthesized 4-(4-bromo-1-naphthyl)-5, 6-dihydro-6(substituted phenyl)-4H-1, 3-oxazine-2-amines by disc-diffusion technique was presented in **Table 2**.

Table 2: Antibacterial activities of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6(substituted phenyl)-4H-1,3-oxazine-2-amines.

Entry	X	<i>E-coli</i>	<i>S.aures</i>	<i>P.aerugin</i> <i>osa</i>	<i>Klebsiella</i>	<i>P.vulgaris</i>	<i>E. faecalis</i>
1	H	+	±	+	±	++	++
2	3-NH ₂	±	++	+	++	±	±
3	4-NH ₂	±	++	+	++	±	±
4	3-Br	++	+	++	++	+	++
5	3-Cl	++	±	++	+	+	+
6	4-Cl	++	+	++	+	++	++
7	4-N(CH ₃) ₂	±	++	+	++	±	++
8	4-OH	±	+	+	±	±	+
9	4-OCH ₃	+	++	±	±	++	±
10	4-CH ₃	+	+	±	±	++	±

11	2-NO ₂	±	+	+	+	±	±
12	3-NO ₂	±	+	+	±	+	±
13	4-NO ₂	+	±	±	+	+	±

Disc size: 6.35 mm; Duration: 24-45 h; Standard: Ampicillin (30-33 mm) and Streptomycin(20-25 mm); Control: Methanol; ---: No activities; ±: Active(8-12 mm); +: Moderately active(13-19 mm); ++: Active(20-24 mm).

Antifungal activity

The measured antifungal activity of synthesized 4-(4-bromo-1-naphthyl)-5,6-dihydro-6(substituted phenyl)-4H-1,3-oxazine-2-amines by disc diffusion technique was presented in Table 3.

Table 3: Antifungal activities of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6(substituted phenyl)-4H-1,3-oxazine-2-amines.

Entry	X	Disc diffusion technique (250 µg/mL)		Drug dilution method (50 µg/mL)	
		<i>Candida albicans</i>		<i>Penicillium</i>	<i>Aspergillus niger</i>
1	H	±		±	+
2	3-NH ₂	+		+	+
3	4-NH ₂	+		+	+
4	3-Br	±		±	+
5	3-Cl	++		++	+
6	4-Cl	+		++	+
7	4-N(CH ₃) ₂	+		±	±
8	4-OH	++		---	++
9	4-OCH ₃	+		+	+
10	4-CH ₃	+		+	---
11	2-NO ₂	++		++	++
12	3-NO ₂	++		++	++
13	4-NO ₂	++		++	++

Standard: Griseofulvin and Gentamycin; Duration: 72 h; Control: Methanol; Medium: Potato dextrose agar; ++: No fungal colony; +: One fungal colony; ±: Two-three fungal colonies; ---: Heavy fungal colony.

DISCUSSION

Synthesis

In our organic chemistry research laboratory, attempts made for synthesizing oxazine derivatives by solvent-free cyclization of chalcones possess electron withdrawing as well as electron donating group as substituents and urea in the presence of various solid catalysts including fly-ash:H₂SO₄ using microwave irradiation. Hence the authors have synthesized some substituted 1,3-oxazine-2-amines derivatives by the cyclization of 2 mmole of 4-bromo-1-naphthyl chalcones, 2 mmole of urea and 0.4g of fly-ash:H₂SO₄ catalyst under microwave irradiation at 550W for 4-6 minutes (Samsung Grill, GW73BD Microwave oven, 230V A/c, 50Hz, 2450Hz, 100-750W (IEC-705), (Scheme 1). During the course of this reaction fly-ash:H₂SO₄ catalyzes cyclization between chalcones and urea followed by rearrangement gave the 1, 3-oxazine-2-amines. The yields of the oxazine in this reaction are more than 80%. The chalcone containing an electron donating substituent (OCH₃) gave higher yields than electron-withdrawing (halogens, NO₂) substituents. Further, we have

investigated this cyclization reaction with equi-molar quantities of the styryl 4-bromo-1-naphthyl ketone (entry 1) with urea under the same condition as above. In this reaction, the obtained yield was 90%. The effect of a catalyst on this reaction was studied by varying the catalyst quantity from 0.1 g to 1 g. As the catalyst quantity is increased from 0.1 g to 1 g, the percentage of yield of product is increased from 84 to 90%. The synthesized 4-(4-bromo-1-naphthyl)-5,6-dihydro-6(substitutedphenyl)-4H-1,3-oxazine-2-amines were characterized by their physical constants and spectroscopic data.

Infrared spectra

The synthesized 4-(4-bromo-1-naphthyl)-5,6-dihydro-6(substituted phenyl)-4H-1,3-oxazine-2-amines(1-13) were characterized by infrared spectral data. The infrared spectra of the parent compound were shown in (figure 1). The peaks obtained at 3427, 1598 and 1253 cm⁻¹ are corresponding stretches of NH, C=N, and C-O-C groups. For all compounds, the NH stretches obtained at the range of 3227-3567 cm⁻¹. The C=N stretches obtained at the range of 1627-1598 cm⁻¹.

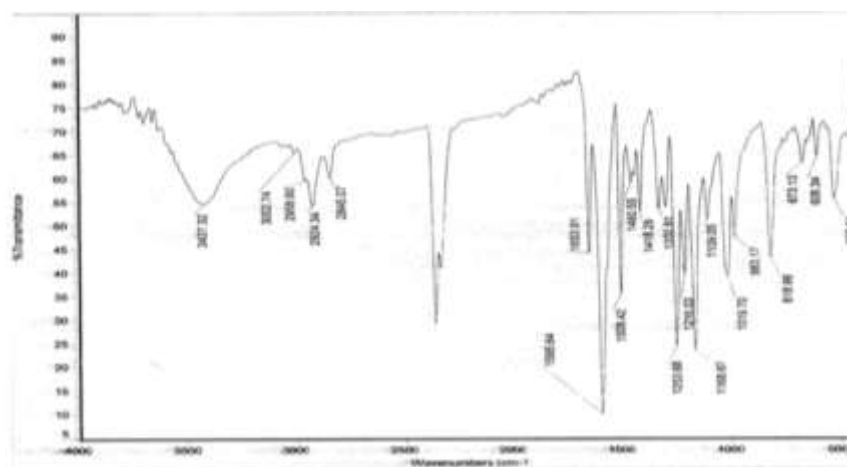


Fig.1: Infrared spectra of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6- phenyl-4H-1,3-oxazine-2-amines(1).

¹H NMR spectra

From proton NMR spectra of synthesized 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-4H-1,3-oxazine-2-amines, the NH₂ proton chemical shifts (δ , ppm) obtained in the range of 6.21-6.71. The chemical shifts (δ , ppm) of H₄ proton gave triplet and it obtained in the range of 4.52-5.39. The proton chemical shifts (δ , ppm) H₅ and H₆ gave doublet of doublet and it obtained in the range of 2.12-2.47 and 2.13-2.73. The proton chemical shifts (δ , ppm) H₆

gave triplet and it falls in the range of 4.35-4.87. These chemical shifts support for the formation of oxazine compounds. The aromatic protons chemical shifts falls in the range of 6.27-8.13 ppm. The parent compound (1) of the oxazine also gave the chemical shifts (δ , ppm) values in this range and this is illustrated in (figure 2). The chemical shifts are 6.71(s, 2H, NH₂), 5.29(t, 1H, H₄), 2.89 (dd, 1H, H₅), 2.13 (dd, 1H, H₆), 4.78(t, 1H, H₆), 7.03-8.13(m, 11H, Ar-H) ppm.

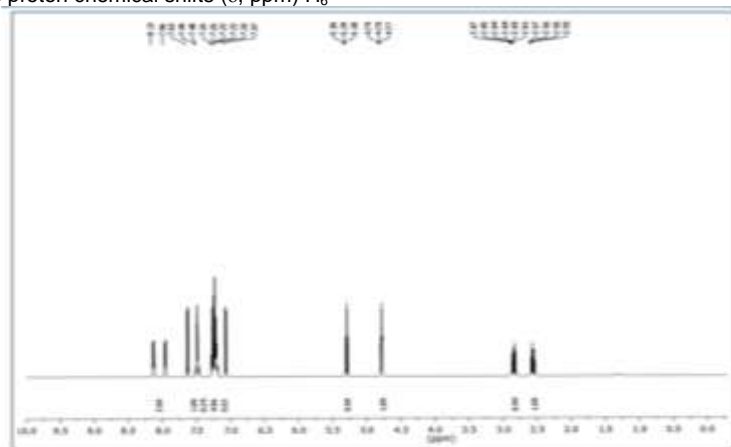


Fig. 2: The ¹H NMR spectra of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-phenyl-4H-1,3-oxazine-2-amines.

¹³C NMR spectra

The ¹³C NMR spectral data reveals that the chemical shifts of the ring carbons C₂, C₄, C₅, C₆ and aromatic carbons falls in the range 154.25-155.70, 49.75-51.76, 39.83-47.80, 66.29-73.19 and 114.23-145.90 ppm. The ¹³C NMR spectra of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-phenyl-4H-1,3-oxazine-2-amine was shown in (figure 3) the corresponding chemical shifts

are 155.19(C₂), 49.75(C₄), 39.83(C₅), 73.19(C₆), 124.39-140.90(Ar-C). These data are supported for the formation of oxazines.

Further, these oxazines were confirmed by mass spectra. The mass spectral fragments gave the corresponding parent and M⁺¹ ion peaks. Similarly, all compounds gave the isotopic peaks of M⁺² and M⁺⁴ also obtained with a respective mass of fragments.

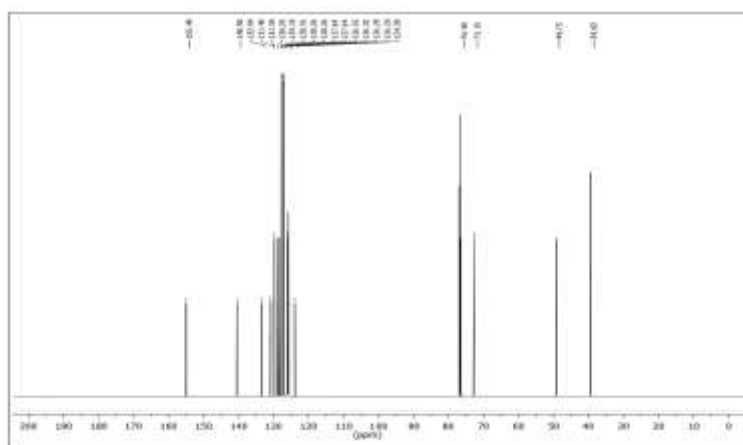


Fig.3: The ¹³C NMR spectra of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-phenyl-4H-1,3-oxazine-2-amine(1)

Antibacterial activity assay

From the observed antibacterial activities of the synthesized oxazines by means of measurement of mm of zone of inhibition, the oxazine-2-amines 4-6 showed maximum zone of inhibition against *Escherichia coli*, with greater than 20 mm compared to the oxazines 1, 9, 10 and 13 and they are moderately active within 13-19 mm of zone of inhibition. Compounds 2, 3, 7, 8, 11 and 13 were active within 8-12 mm of the zone of inhibition. The oxazine-2-amine derivatives 2, 3, 7 and 9 were found to be effective against *S. aureus* within 20-24 mm of the zone of inhibition. Compounds 4, 6 and 8 are moderately active within 13-19 mm of the zone of inhibition. The oxazines 1, 5 and 13 were active within 8-12 mm of the zone of inhibition. The oxazine derivatives 4-6 were more active against *Pseudomonas* showing greater than 20 mm zone of inhibition and the other derivatives 1-3, 7, 8, 11 and 12 were showed the zone of inhibitions between

12-19 mm. The oxazines 9, 10 and 13 were showed active within 8-12 mm of the zone of inhibition. The oxazine-2-amines 2-4 and 7 were effective within 20-24 mm of the zone of inhibition against *Klebsiella*. The compounds 5, 6 and 13 were active with 12-12 mm zone of inhibition while the other oxazine showed a moderate activity. The compounds 1, 6, 9 and 10 were active when it is screened against *P. vulgaris* and the other compounds are less effective. The oxazines 1, 4, 6 and 7 showed activities against *E-faecalis* when they are screened with 20-24 mm zone of inhibition.

Antifungal activity assay

The study of antifungal activities of all 4-(1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines against *C. albicans*, showed that the five compounds 5, 8, 11-13 are effective with 20 mm as the zone of inhibition in 250 μ g/ disc while oxazine-2-amines 2, 3, 6, 7, 9 and 10 were active with 13-19 mm zone of

inhibition and the compound 1 and 4 were the least active with 8-12 mm zone of inhibitions. The oxazine derivatives 6, 11-13 were more visible against *Penicillium* species, in the development of the fungal colony and 2-3 colonies are recorded for the compounds 1-3, 5, 9 and 10. The inhibition of oxazine-2-amines against *A.niger* was more active in two compounds 5, 8, 11-13. The compounds 2-4, 6 and 9 were active with 8-12 mm zone of inhibitions. The presence of amino-, dimethylamino-, methoxy-, methyl-, dimethyl-, chloro-, bromo- and nitro- substituents are responsible for antimicrobial activities of 4-(4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines.

CONCLUSIONS

Some (4-bromo-1-naphthyl)-5,6-dihydro-6-(substituted phenyl)-oxazine-2-amines have been synthesized by solid fly-ash: H₂SO₄ catalyzed solvent-free cyclization of aryl chalcones and urea using microwave irradiation under solvent-free conditions. The yields of these oxazines were more than 80%. This synthetic methodology offers solvent-free cyclization, non-hazardous, easy-workup procedure, shorter reaction time and better yields. These oxazine derivatives were characterized by their physical constants and spectral data. The antimicrobial activities of these 4-(4-bromo-1-naphthyl)-5, 6-dihydro-6-(substituted phenyl)-oxazine-2-amines have been evaluated. The compounds possess amino, dimethylamino methoxy, methyl, dimethyl, chloro, bromo and nitro substituents were showed antimicrobial activities at 20-24, 13-19 and 8-12 mm of the zone of inhibitions.

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